

ORP Sensor

(Order Code ORP-BTA)



The ORP Sensor measures the ability of a solution to act as an oxidizing agent or reducing agent. ORP stands for oxidation-reduction potential. For example, ORP electrodes are often used to measure the oxidizing ability of chlorine in swimming pools, or to determine when the equivalence point has been reached in an oxidation-reduction reaction.

Compatible Software

See www.vernier.com/manuals/orp-bta for a list of software compatible with the ORP Sensor.

Quick Start

1. Plug the sensor into the interface (LabQuest 3, LabQuest Mini, etc.).
2. Connect the interface to your device.
 - If using USB, connect to the USB port on your computer.
 - If using Bluetooth® wireless technology, click your interface type and then select your device.
3. Prepare for data collection:
 - Vernier Graphical Analysis®: Launch the app, if necessary, and click Sensor Data Collection.
 - LabQuest® App: Choose New from the File menu.

The software will identify the sensor and load a default data-collection setup. You are now ready to collect data.

Note: Do not completely submerge the sensor. The handle is not waterproof.

Need Additional Information?

Visit the following link:

www.vernier.com/start-lq-sensor

Note: Vernier products are for educational use only.

Calibrating the Sensor

Optional Calibration Procedure

In most experiments done with an ORP Sensor the precise potential in mV is not critical; rather, the large change in potential is the most important factor. As a result, you should not have to perform a new calibration when using the ORP Sensor for most experiments. You can simply use the appropriate calibration that is stored with this auto-ID sensor.

If you are doing water quality testing or performing a chemistry experiment that requires a very accurate calibration, you will need two commercial ORP standards. Using these standards, perform the following calibration, using the 2-point calibration option in all Vernier data-collection programs (this calibration assumes you have two ORP calibration standards, one at 100 mV, another at 300 mV):

- For the first calibration point, rinse the tip of the electrode with distilled water, and place the electrode into the first standard. When the voltage reading displayed by the data-collection program stabilizes, enter the mV value of the first ORP standard (e.g., 100).
- For the second calibration point, remove the electrode from the first standard, rinse it with distilled water, and place it into the second standard. When the voltage stabilizes, enter the mV reading of the second standard (e.g., 300).
- Rinse the electrode with distilled water and place it into the sample. You are now ready to take measurements with the calibrated ORP Sensor.

When you are finished making measurements, rinse the electrode with distilled water. Slide the cap onto the electrode body, and then screw the cap onto the storage bottle so the tip of the electrode is immersed in the storage solution.

Specifications

ORP Electrode

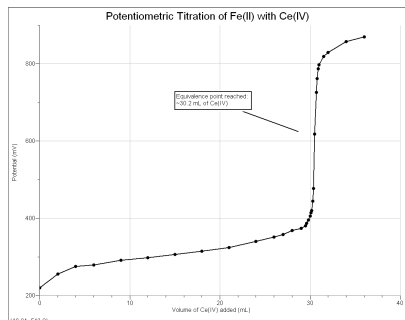
Type	Sealed, gel-filled, epoxy body, Ag/AgCl reference
Storage solution	pH-4/KCl solution (10 g KCl in 100 mL buffer pH-4 solution)
Cable	1 meter coaxial cable with BNC connector
Temperature Range	0 to 60°C 12mm OD Impedance: ~20 kΩ at 25°C
ORP element	99% pure platinum band sealed on a glass stem

Electrode Amplifier (included with each ORP Sensor)

Calibration	slope: 466.875 intercept: -559.793
Resolution	0.5 mV
Power	7 mA @ 5VDC
Input Range	-450 to 1100 mV

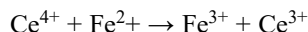
Suggested Experiments

The ORP Sensor can be used to perform a potentiometric titration. This is a fairly common experiment in AP Chemistry or college general chemistry classes. When a redox titration just exceeds its equivalence point volume, the potential measured by an ORP electrode will increase rapidly (if there is an excess of oxidizing agent) or decrease rapidly (with excess reducing agent), as seen in the graph below.

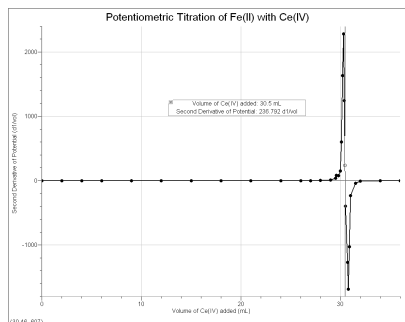


ORP titration of Fe^{2+} solution with Ce^{4+}

In the reaction for the titration curve shown previously



a solution containing Fe^{2+} of unknown concentration is titrated with an oxidizing agent, ~ 0.1 M Ce^{4+} standard solution (from $(NH_4)_2Ce(NO_3)_6$). When the equivalence point is reached, and excess Ce^{4+} is added, a large increase in potential results. By examining these data, or performing a second derivative (also shown) or Gran plot, the equivalence point of the titration can easily be determined. With Vernier Graphical Analysis 4 software, the experiment can be done in one of two ways: using the ORP Sensor in Events with Entry mode (where buret volumes are manually entered) or using the Vernier Drop Counter to measure titrant volumes.



Second derivative plot for the titration of Fe^{2+} solution with Ce^{4+}

Care and Maintenance

Do not completely submerge the sensor. The handle is not waterproof.

Do not wrap the cable tightly around the sensor for storage. Repeatedly doing so can irreparably damage the wires and is not covered under warranty.

How the Sensor Works

The electrode has two components: a *measuring* half cell comprised of platinum metal immersed in the solution in which the redox reaction is taking place, and a *reference* half cell (sealed gel-filled Ag/AgCl) to which the platinum half cell is referenced.

The Vernier ORP can measure redox potential in the range of -450 to $+1100$ mV. Readings toward the positive region of this range indicate a strong oxidizing agent, while readings toward the negative region indicate a strong reducing agent.

Troubleshooting

For best results when using the ORP sensor with a drop counter, add drops at a rate of one per two seconds.

For troubleshooting and FAQs, see www.vernier.com/til/1437

Repair Information

If you have followed the troubleshooting steps and are still having trouble with your ORP Sensor, contact Vernier Technical Support at support@vernier.com or call 888-837-6437. Support specialists will work with you to determine if the unit needs to be sent in for repair. At that time, a Return Merchandise Authorization (RMA) number will be issued and instructions will be communicated on how to return the unit for repair.

Warranty

Warranty information for this product can be found on the Support tab at www.vernier.com/orp-bta

General warranty information can be found at www.vernier.com/warranty



Vernier Science Education

13979 SW Millikan Way • Beaverton, OR 97005-2886

Toll Free (888) 837-6437 • (503) 277-2299 • Fax (503) 277-2440

info@vernier.com • www.vernier.com

Rev. 8/6/2024

Vernier Graphical Analysis, LabQuest, LabQuest Mini, and other marks shown are our trademarks or registered trademarks in the United States.

All other marks not owned by us that appear herein are the property of their respective owners, who may or may not be affiliated with, connected to, or sponsored by us.

